# POZNAN UNIVERSITY OF TECHNOLOGY



EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS) pl. M. Skłodowskiej-Curie 5, 60-965 Poznań

# **COURSE DESCRIPTION CARD - SYLLABUS**

Course name				
Fundamentals of electrochemical technology				
Course				
Field of study		Year/Semester		
Chemical and process engineering		3/5		
Area of study (specialization)		Profile of study		
		general academic		
Level of study		Course offered in		
First-cycle studies		Polish		
Form of study		Requirements		
full-time		compulsory		
Number of hours				
Lecture	Laboratory classes	Other (e.g. online)		
30	30			
Tutorials	Projects/seminars			
Number of credit points				

## 5

## Lecturers

Responsible for the course/lecturer: dr hab. Piotr Krawczyk, prof. PP

Responsible for the course/lecturer:

## **Prerequisites**

Student has a ordered knowledge of mathematics and physical chemistry and he also has ability to use the basic techniques in a laboratory scale.

## **Course objective**

The aim of the course is to familiarize students with an overview of technical electrochemistry methods and develop skills for their practical application.

## **Course-related learning outcomes**

#### Knowledge

1. The knowledge in the field of basics of electrochemical processes -[ K W03, K W04],

2. The knowledge in the field of various electrochemical technologies -[K W13, K W15],

## 3. The knowledge in the field of related fields –[K\_W12].

#### Skills

1. The student has the ability to selection of measurement techniques -[K\_U15, K\_U16],



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2. The student has the ability to use specialized vocabulary in English –[K\_U01, K\_U02].

#### Social competences

1. The student understands the need for self-study and improvement of their professional competence –[K\_K01],

2. Student can act and cooperate in the group accepting different roles –[K\_K04].

## Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

Laboratory assessment on the basis of the current work during the laboratory and the written tests.

The written exam.

### **Programme content**

- 1. The principles of electrochemical processes.
- 2. Electrodes balances.
- 3. The kinetics of electrode processes.
- 4. The selected electrochemical processes.
- 5. The processes based on the electrochemical processes.

6. Construction solutions of electrochemical reactors and their influnce on the course of electrochemcial processes.

## **Teaching methods**

Lecture, problem lecture, explanation, didactic discussion, classes, project method, laboratory exercises

## Bibliography

Basic

1. A. Kisza – Elektrochemia cz. I i II (Jonika i Elektrodyka) WNT, W-wa, 2001,

- 2. R. Dylewski, W. Gniot, M. Gonet, Elektrochemia przemysłowa, Wyd. Politechniki Śląskiej, 1999,
- 3. A. Czerwiński, Ogniwa, akumulatory, baterie, WNT, W-wa, 1999,
- 4. C. G. Zoski praca zb., Handbook of Electrochemistry, Elsevier, 2007,
- 5. A. Ciszewski, Technologia chemiczna. Procesy elektrochemiczne, Wyd. Politechniki Poznańskiej, 2008.

## Additional

- 1. A.V. da Rosa, Fundamentals of Renewable Energy Processes, Elsevier/Academic Press, 1990,
- 2. H. Scholl, T. Błaszczyk, P. Krzyczmonik, Elektrochemia, Wyd. Uniwersytetu Łódzkiego, 1998.

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# Breakdown of average student's workload

	Hours	ECTS
Total workload	140	5,0
Classes requiring direct contact with the teacher	70	2,5
Student's own work (literature studies, preparation for laboratory	70	2,5
classes/tutorials, preparation for tests/exam, project preparation) <sup>1</sup>		

<sup>&</sup>lt;sup>1</sup> delete or add other activities as appropriate